

A stable isotope aridity index for terrestrial environments

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We use the oxygen isotopic composition of tooth enamel from multiple mammalian taxa across eastern Africa to present a proxy for aridity. Here we report tooth enamel $\delta^{18}\text{O}$ values of 14 species from 18 locations and classify them according to their isotopic sensitivity to environmental aridity. The species are placed into two groups, evaporation sensitive (ES) and evaporation insensitive (EI). Tooth enamel $\delta^{18}\text{O}$ values of ES animals increase with aridity, whereas the tooth enamel $\delta^{18}\text{O}$ values of EI animals track local meteoric water $\delta^{18}\text{O}$ values, demonstrating that bioapatite $\delta^{18}\text{O}$ values of animals with different behaviors and physiologies record different aspects of the same environment. The enrichment between tooth enamel $\delta^{18}\text{O}$ values of ES and EI animals records the degree of ^{18}O enrichment between evaporated water (ingested water or body water) and source water, which increases with environmental aridity. Recognition of the ES–EI distinction creates the opportunity to use the ^{18}O composition of bioapatite as an index of terrestrial aridity.

bioapatite | East Africa | oxygen-18 | mammals | water use

Terrestrial responses to major climate changes, such as glaciations, orogenic events, and shifts in ocean circulation, are often characterized in terms of water availability or aridity (1–3). Although aridity proxies exist for different terrestrial settings (4–6), they are not applicable in every circumstance and additional proxies must be developed for further study of terrestrial environmental change. The ^{18}O composition of bioapatite has been used as a proxy for rainfall $\delta^{18}\text{O}$ and seasonality of past environments (7, 8), but its utility in paleoenvironmental problems is limited by the complexity of climatic, environmental, physiological, and behavioral variables that influence bioapatite $\delta^{18}\text{O}$ values. The correlation between bioapatite $\delta^{18}\text{O}$ values and both meteoric water $\delta^{18}\text{O}$ values and relative humidity demonstrate that animals have different isotopic responses to environmental change (7, 9, 10). In this study, we present tooth enamel $\delta^{18}\text{O}$ data of 14 mammal species sampled from 18 locations in eastern Africa, which represent a gradient in environmental aridity. This data set shows that the tooth enamel $\delta^{18}\text{O}$ values of some species vary with aridity whereas those of other species track the ^{18}O composition of meteoric water. We use the different isotopic responses of the sampled species as the empirical basis for an aridity proxy.

Results

There is a marked increase in water deficit (WD) from the closed canopy Ituri Forest in the Democratic Republic of Congo to the arid shrublands of Lake Turkana in northern Kenya (Table 1, which is published as supporting information on the PNAS web site). WD in these regions negatively correlates to mean annual relative humidity (RH) ($P < 0.01$) and is used as a measure of aridity. Although previous studies compare bioapatite $\delta^{18}\text{O}$ values with RH (6, 9, 10), RH data are not used here because they are not available for all study locations. Meteoric water $\delta^{18}\text{O}$ ($\delta^{18}\text{O}_{\text{mw}}$) values at these localities average -3.1‰ (SE, 1.2‰) and do not vary significantly with WD ($P = 0.476$). We use this

average $\delta^{18}\text{O}_{\text{mw}}$ value for localities where water isotope data are not available.

The isotopic enrichment between tooth enamel and meteoric water ($\varepsilon_{\text{enamel-mw}}$), where $\varepsilon_{\text{A-B}} = ((R_{\text{A}}/R_{\text{B}}) - 1) \times 1,000$ is plotted as a function of WD (Figs. 1 and 2). The use of an enrichment factor, ε , enables tooth enamel $\delta^{18}\text{O}$ ($\delta^{18}\text{O}_{\text{enamel}}$) values to be compared between sites, regardless of changes in $\delta^{18}\text{O}_{\text{mw}}$ values. $\varepsilon_{\text{enamel-mw}}$ for hippopotamus (*Hippopotamus amphibius*), bush pig (*Potamochoerus larvatus*), elephant (*Loxodonta africana*), rhinoceros (*Diceros bicornis*), warthog (*Phacochoerus africanus*), zebra (*Equus burchelli*), impala (*Aepyceros melampus*), and baboon (*Papio anubis*), is not sensitive to changes in WD (P values range from 0.14 to 0.91) (Fig. 1). Baboons from Mpala are the only mixed ($\text{C}_3\text{--C}_4$) feeding baboons (otherwise C_3 browsers), and when not considered the baboon regression has a P value < 0.001 instead of 0.14. $\varepsilon_{\text{enamel-mw}}$ of a second group, giraffids (*Giraffa camelopardalis* and *Okapia johnstoni*), oryx (*Oryx beisa*), dikdik (*Madoqua kirkii*), Grant's gazelle (*Gazella granti*), and buffalo (*Syncerus caffer*), increases with WD (P values range from < 0.001 to 0.08) (Fig. 2 *a–e*). The regression coefficients (slopes) of the second group do not vary significantly from each other ($P > 0.1$, F test).

Discussion

The Evaporation Sensitive (ES)–Evaporation Insensitive (EI) Distinction. The regressions between WD and $\varepsilon_{\text{enamel-mw}}$ show that aridity affects $\delta^{18}\text{O}_{\text{enamel}}$ values of some animals and not others. We assign the ES classification to those animals (giraffids, dikdik, and oryx) for which $\varepsilon_{\text{enamel-mw}}$ is sensitive to WD ($P < 0.05$). $\varepsilon_{\text{enamel-mw}}$ values of EI animals (hippopotamus, warthog, elephant, rhinoceros, and zebra) do not vary with aridity ($P > 0.10$). Although high regression coefficients ($> 10^{-3}$) suggest that the response of $\varepsilon_{\text{enamel-mw}}$ values to WD for baboon, impala, Grant's gazelle, and buffalo, is similar to that for ES taxa, these animals are placed in neither category because the regressions are not significant at the 0.05 level. ANOVA comparisons using a post hoc Scheffé test show that, within a single location, $\delta^{18}\text{O}_{\text{enamel}}$ values of EI animals (hippopotamus, warthog, elephant/rhinoceros, and zebra) form significantly different groups ($P < 0.05$), whereas $\delta^{18}\text{O}_{\text{enamel}}$ values of the ES animals (giraffids, dikdik, and oryx) do not vary significantly from each other ($P > 0.05$). Bushpig $\delta^{18}\text{O}_{\text{enamel}}$ values are not included in the ES–EI classification because the data represent a very small range in WD.

The evaporative enrichment of ^{18}O in ingested water and in body water has a strong influence on $\delta^{18}\text{O}$ and δD values of both mammalian and avian body tissues, and it is a likely explanation for the sensitivity of ES $\varepsilon_{\text{enamel-mw}}$ values to aridity (9–12). The ^{18}O composition of leaf, surface, and body waters increase with greater evaporation, but the specific isotopic response of waters

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Abbreviations: WD, water deficit; RH, relative humidity; ES, evaporation sensitive; EI, evaporation insensitive.

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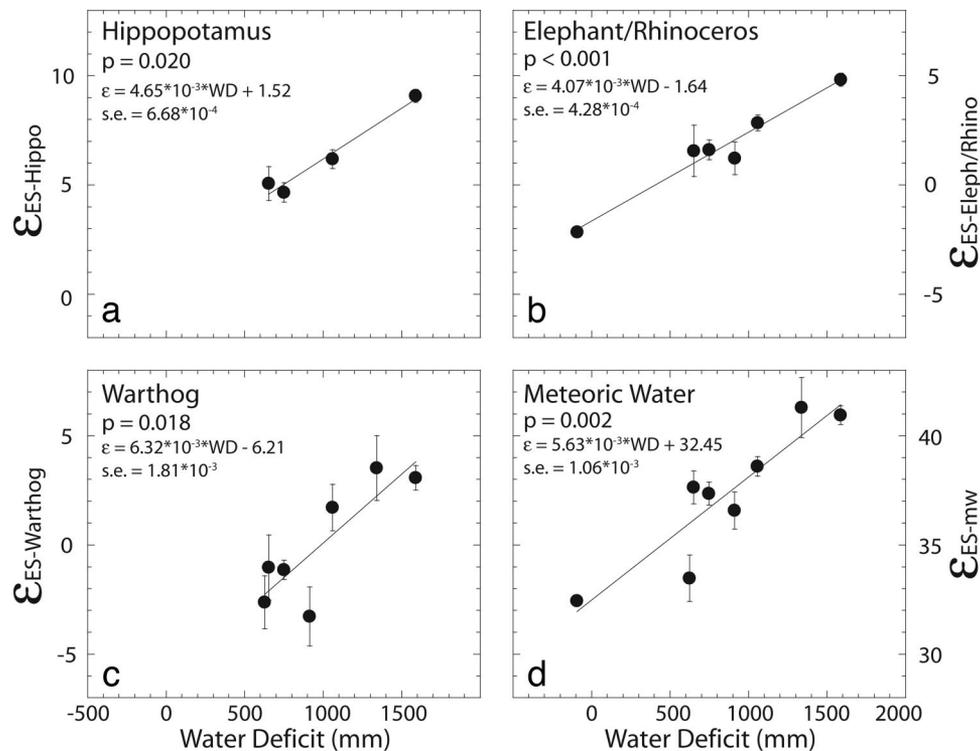


Fig. 3. The isotopic enrichment (ϵ) between ES $\delta^{18}\text{O}_{\text{enamel}}$ values (averaged values of all ES specimens at each site) and $\delta^{18}\text{O}_{\text{enamel}}$ values of hippopotamus (a), elephant/rhinoceros (b), warthog (c), and meteoric water (d) is plotted with water deficit. ES and EI taxa are grouped based on ANOVA results presented in text. Linear regressions, P values, and the standard error (s.e.) on the regression coefficient are reported for each plot. Error bars represent the standard error associated with each ϵ value.

within one sedimentological stratum or well correlated unit. To reduce the possibility that animals accessed evaporated surface waters, fossils from lacustrine sediments should be avoided unless it can be demonstrated that the lake water was not enriched in ^{18}O relative to meteoric water.

With the above criteria met, the aridity index can be used to convert differences in $\epsilon_{\text{ES-EI}}$ between sites to differences in WD. The use of $\epsilon_{\text{ES-EI}}$ and not absolute $\delta^{18}\text{O}_{\text{enamel}}$ values eliminates the need to know taxon-specific fractionation factors between environmental water and body water $\delta^{18}\text{O}$ values, and it controls for changes in $\delta^{18}\text{O}_{\text{mw}}$ values that are independent of aridity.

Conclusion

The empirical relationships presented here lay the foundation for a simple approach that relates $\delta^{18}\text{O}_{\text{enamel}}$ values to aridity when it is not practical to model the flux of $^{18}\text{O}/^{16}\text{O}$ through an animal. The aridity index is a general relationship between aridity and $\delta^{18}\text{O}_{\text{enamel}}$ values that is built on the distinction between ES and EI animals, wherein ES $\delta^{18}\text{O}_{\text{enamel}}$ values track evaporation of leaf or body water and EI $\delta^{18}\text{O}_{\text{enamel}}$ values track the ^{18}O composition of meteoric water (drinking, stem, and root water). The isotopic enrichment between $\delta^{18}\text{O}_{\text{enamel}}$ values of ES and EI animals, $\epsilon_{\text{ES-EI}}$, increases linearly with water deficit along a consistent slope, the aridity index, which can be used to calculate relative differences in water deficit through time or across space. The development of the aridity index with data from modern eastern African mammals makes the aridity index most immediately applicable to East African paleoenvironments, where aridity is considered to have influenced human evolution (27). However, the ES–EI distinction is a broadly applicable concept that can be used to evaluate animal water-use strategies in terrestrial systems where the ^{18}O composition of biapatite is unaltered.

Materials and Methods

In the past 10 years, modern mammal teeth from animals living within the past 45 years have been sampled from museum collections and field locations across eastern Africa for stable isotopic analysis. For this study, a subset of the larger tooth enamel isotopic data set (493 teeth of 1,000) was used to place better control on water sources accessed by animals (Table 2, which is published as supporting information on the PNAS web site). Tooth enamel $\delta^{18}\text{O}$ ($\delta^{18}\text{O}_{\text{enamel}}$) values were only included in this subset if they were from animals with $^{18}\text{O}_{\text{enamel}}$ results from four or more locations. Of these animals, $^{18}\text{O}_{\text{enamel}}$ data were excluded if (i) they were from locations with large isotopic gradients (e.g., mountainous regions) or (ii) they were from animals known to drink evaporated water enriched in ^{18}O with respect to local meteoric water.

Waters used to approximate $\delta^{18}\text{O}_{\text{mw}}$ values were sampled from sources with little evaporation (e.g., rivers with small catchments, springs, and shallow wells). Environmental aridity is characterized in terms of WD, the difference between potential evapotranspiration (PET) and mean annual precipitation (MAP), ($\text{WD} = \text{PET} - \text{MAP}$) using available data from references (28–31) and from N. Georgiadis (personal communication). PET was calculated from mean annual temperature and latitude (32).

Enamel was separated from dentine and powdered with a diamond or carbide bit following the long-axis of the tooth. Where multiple samples from each tooth were taken, the average value was used and considered as one sample. Enamel powders were pretreated by standard procedures in preparation for stable isotope analysis of the carbonate component of tooth enamel (33, 34). Waters were prepared for isotopic analysis by equilibration with CO_2 . Isotope ratios were measured on a Finnigan MAT 252 mass spectrometer at Stable Isotope Ratio Facility for Environmental Research, University of Utah. All isotopic ratios

are reported in δ -notation, where $\delta = ((R_{\text{sample}}/R_{\text{standard}}) - 1) \times 1,000$ and R is $^{18}\text{O}/^{16}\text{O}$. δ values of tooth enamel are reported in reference to Vienna PeeDee Belemnite (VPDB), and waters are reported in reference to Vienna standard mean ocean water (VSMOW). For calculations in which it was necessary to convert values in the VPDB scale to the VSMOW scale, the following relationship was used: $\delta^{18}\text{O}_{\text{VSMOW}} = 1.3086 \times \delta^{18}\text{O}_{\text{VPDB}} + 30.86$ (35). The enrichment between related isotopic values is reported as ϵ , where $\epsilon_{\text{A-B}} = ((R_{\text{A}}/R_{\text{B}}) - 1) \times 1,000$.

For sites where data were available, $\delta^{18}\text{O}_{\text{lw}}$ values were calculated by using a modified version of the 1965 Craig and Gordon evaporative enrichment model (13, 16, 36). To calculate $\delta^{18}\text{O}_{\text{lw}}$ values, mean annual temperature, mean annual RH (0600 h), and water $\delta^{18}\text{O}$ values were used, and it was assumed

that atmospheric humidity ^{18}O composition is 9‰ lower than measured $\delta^{18}\text{O}_{\text{water}}$ values, there is no temperature difference between the leaf and air, stomatal conductance = $0.3 \text{ m}^{-2}\cdot\text{s}^{-1}$, and boundary layer conductance = $1 \text{ m}^{-2}\cdot\text{s}^{-1}$. Statistics were computed with the statistical program JMP 4 (SAS Institute) and SYSTAT 10 (SPSS), or using methods described in Sokal and Rohlf (37).

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1. Sirocco, F., Seelos, K., Schaber, K., Rein, B., Dreher, F., Diehl, M., Lehne, R., Jager, K., Krbetschek, M. & Degering, D. (2005) *Nature* **436**, 833–836.
2. Dettman, D. L., Fang, X. M., Garzzone, C. N. & Li, J. J. (2003) *Earth Planet. Sci. Lett.* **214**, 267–277.
3. Cane, M. A. & Molnar, P. (2001) *Nature* **411**, 157–162.
4. Wilf, P. (2000) *Geol. Soc. Am. Bull.* **112**, 292–307.
5. Retallack, G. J. (2005) *Geology* **33**, 333–336.
6. Cormie, A. B., Luz, B. & Schwarcz, H. P. (1994) *Geochim. Cosmochim. Acta* **58**, 3439–3449.
7. Longinelli, A. (1984) *Geochim. Cosmochim. Acta* **48**, 385–390.
8. Fricke, H. C., Clyde, W. C. & O'Neil, J. R. (1998) *Geochim. Cosmochim. Acta* **62**, 1839–1850.
9. Luz, B., Cormie, A. B. & Schwarcz, H. P. (1990) *Geochim. Cosmochim. Acta* **54**, 1723–1728.
10. Ayliffe, L. K. & Chivas, A. R. (1990) *Geochim. Cosmochim. Acta* **54**, 2603–2609.
11. McKechnie, A. E., Wolf, B. O. & del Rio, C. M. (2004) *Oecologia* **140**, 191–200.
12. Hobson, K. A., Atwell, L. & Wassenaar, L. I. (1999) *Proc. Natl. Acad. Sci. USA* **96**, 8003–8006.
13. Craig, H. & Gordon, L. I. (1965) in *Stable Isotopes in Oceanographic Studies and Paleotemperatures* (V. Lischi, Pisa), pp. 9–130.
14. Gonfiantini, R., Gratziu, S. & Tongiorgi, E. (1965) in *Use of Isotopes and Radiation in Soil-Plant Nutrition Studies*, eds. (International Atomic Energy Agency, Vienna), pp. 405–409.
15. Sternberg, L. d. S. L. (1989) in *Stable Isotopes in Ecological Research*, eds. Rundel, P. W., Ehleringer, J. R. & Nagy, K. A. (Springer, New York), pp. 124–141.
16. Roden, J. S. & Ehleringer, J. R. (1999) *Plant Physiol.* **120**, 1165–1173.
17. Maloiy, G. M. O. (1973) *Proc. R. Soc. London B* **184**, 167–178.
18. Pellew, R. A. (1984) *J. Zool. Soc. London* **202**, 57–81.
19. Taylor, C. R. (1969) *Sci. Am.* **220**, 89–95.
20. Taylor, C. R. (1968) *Nature* **219**, 181–182.
21. Sinclair, A. R. E. (1977) *The African Buffalo* (The University of Chicago Press, Chicago).
22. Western, D. (1975) *E. Afr. Wildlife J.* **13**, 265–286.
23. Rodgers, W. A. (1984) *Mammalia* **48**, 327–350.
24. Schenkel, R. & Schenkel-Hulliger, L. (1969) *Ecology and Behavior of the Black Rhinoceros (*Diceros bicornis* L.)* (Verlag Paul Parey, Hamburg).
25. Lillywhite, H. B. & Stein, B. R. (1987) *J. Zool. (London)* **211**, 727–734.
26. Clementz, M. T. & Koch, P. L. (2001) *Oecologia* **129**, 461–472.
27. deMenocal, P. B. (2004) *Earth Planet. Sci. Lett.* **220**, 3–24.
28. East African Meteorological Department (1975) *Climatological Statistics for East Africa* (East African Meteorological Department, Nairobi, Kenya).
29. East, R. (1984) *Afr. J. Ecol.* **22**, 245–270.
30. Altmann, J., Alberts, S. C., Altmann, S. A. & Roy, S. B. (2002) *Afr. J. Ecol.* **40**, 248–251.
31. Cerling, T. E., Hart, J. A. & Hart, T. B. (2004) *Oecologia* **138**, 5–12.
32. Thornthwaite, C. W. (1948) *Geogr. Rev.* **38**, 55–94.
33. Lee Thorp, J. & van der Merwe, N. J. (1987) *S. Afr. J. Sci.* **83**, 712–715.
34. Koch, P. L., Tuross, N. & Fogel, M. L. (1997) *J. Arch. Sci.* **24**, 417–429.
35. Friedman, I. & O'Neil, J. R. (1977) *Compilation of Stable Isotope Fractionation Factors of Geochemical Interest* (U.S. Geological Survey, Reston, VA) p. 1–12.
36. Cappa, C. D., Hendricks, M. B., DePaolo, D. J. & Cohen, R. C. (2003) *J. Geophys. Res. Atmos.* **108**, doi:10.1029/2003JD003597.
37. Sokal, R. R. & Rohlf, F. J. (1995) in *Biometry* (Freeman, New York).