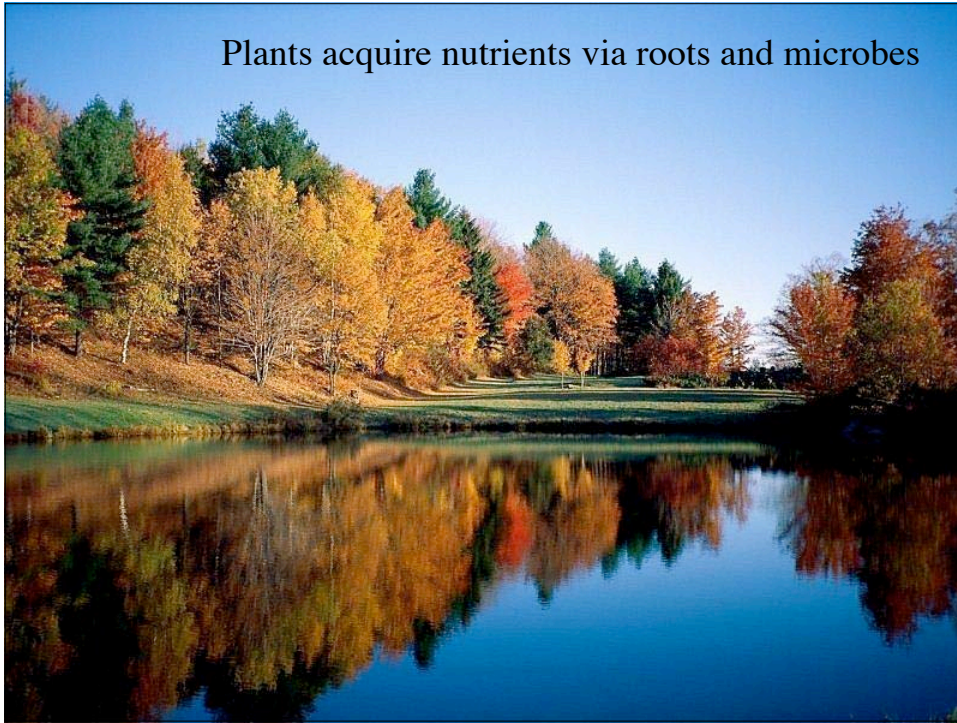
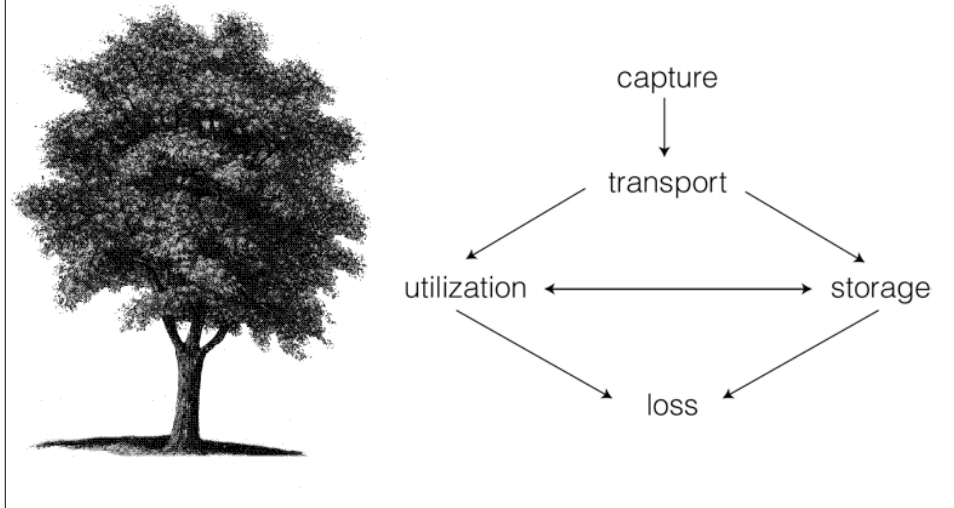
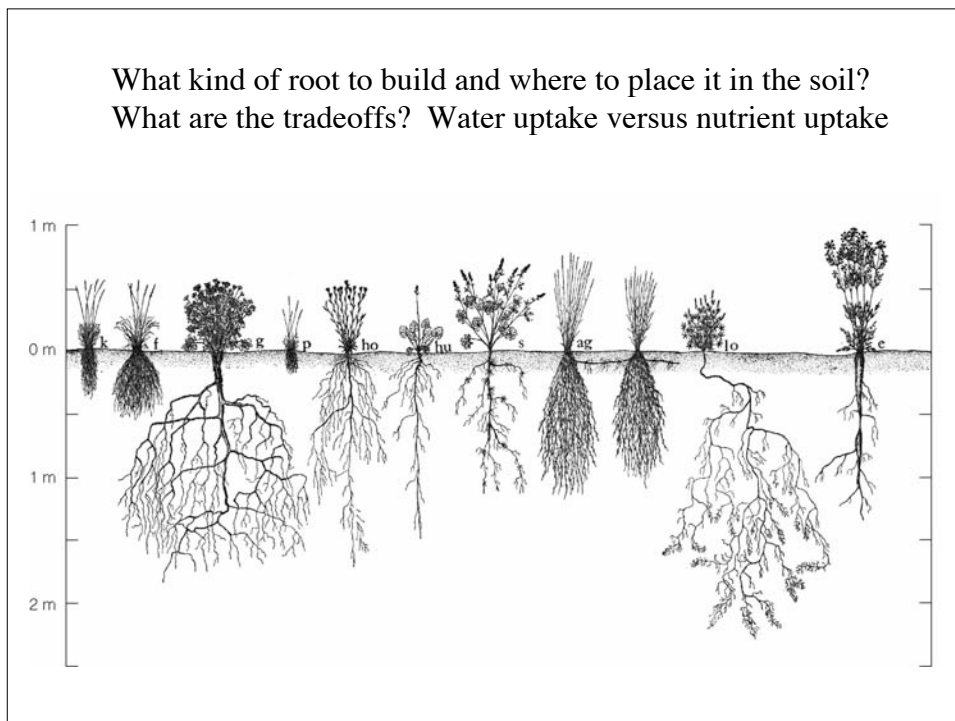
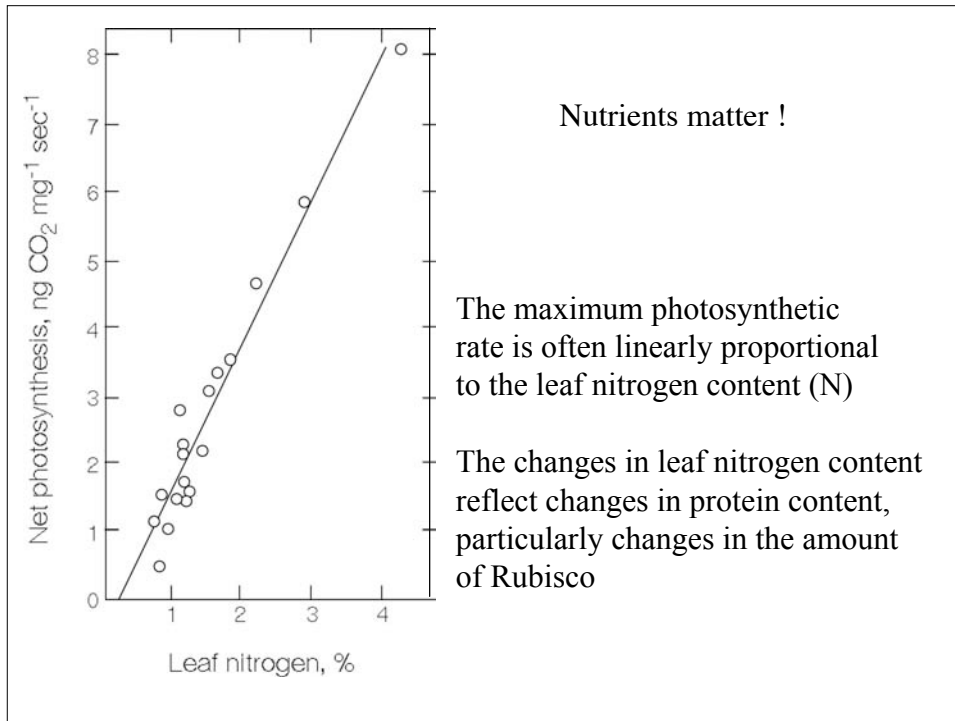


Plants acquire nutrients via roots and microbes





Consideration of whole plant as an integrated system:  
how are resources used, since photosynthesis is  
only part of the story





## How to acquire the nutrients needed for growth?

- Nutrient sources
  - weathering, atmospheric deposition, N-fixation, recycling within the soil (within ecosystem)
- Plant uptake
  - movement of nutrients, acquisition strategies
- Symbioses with microorganisms
  - mycorrhizae, symbiotic N-fixation

Element	Principal form absorbed	Usual concentration in healthy plants (% or ppm of Dry Weight)	Important Functions
<b>Macronutrients</b>			
Carbon (C)	CO <sub>2</sub>	~44%	Component of organic compounds
Oxygen (O)	H <sub>2</sub> O or O <sub>2</sub>	~44%	Component of organic compounds
Hydrogen (H)	H <sub>2</sub> O	~6%	Component of organic compounds
 Nitrogen (N)	NO <sub>3</sub> <sup>-</sup> or NH <sub>4</sub> <sup>+</sup>	1-4%	Component of amino acids, proteins, nucleotides, nucleic acids, chlorophylls, and coenzymes
Potassium (K)	K <sup>+</sup>	0.5-6%	Involved in osmosis and ionic balance and in opening and closing of stomata; activator of many enzymes
Calcium (Ca)	Ca <sup>2+</sup>	0.2-3.5%	Component of cell walls; enzyme cofactor; involved in cellular membrane permeability; component of calmodulin, a regulator of membrane and enzyme activities
 Phosphorus (P)	H <sub>2</sub> PO <sub>4</sub> <sup>-</sup> or HPO <sub>4</sub> <sup>2-</sup>	0.1-0.8%	Component of energy-carrying phosphate compounds (ATP and ADP), nucleic acids, several essential coenzymes, phospholipids
Magnesium (Mg)	Mg <sup>2+</sup>	0.1-0.8%	Part of the chlorophyll molecule; activator of many enzymes
Sulfur (S)	SO <sub>4</sub> <sup>2-</sup>	0.05-1%	Component of some amino acids and proteins and of coenzyme A

(Raven et al. 1992, *Biology of Plants*)

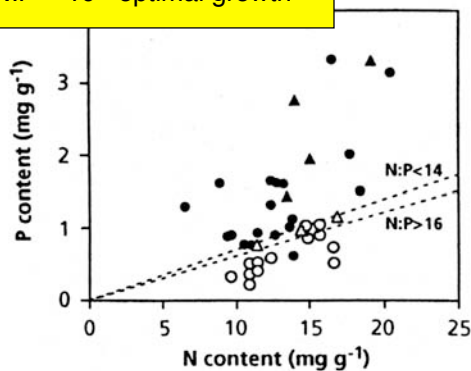
Element	Principal form absorbed	Usual concentration in healthy plants (% or ppm of Dry Weight)	Important Functions
<b>Micronutrients</b>			
Iron (Fe)	Fe <sup>2+</sup> or Fe <sup>3+</sup>	25-300 ppm	Required for chlorophyll synthesis; component of cytochromes and nitrogenase
Chlorine (Cl)	Cl <sup>-</sup>	100-10,000 ppm	Involved in osmosis and ionic balance; probably essential in photosynthetic reactions that produce oxygen
Copper (Cu)	Cu <sup>2+</sup>	4-30 ppm	Activator or component of some enzymes
Manganese (Mn)	Mn <sup>2+</sup>	15-800 ppm	Activator of some enzymes; required for integrity of chloroplast membrane and for oxygen release in photosynthesis
Zinc (Zn)	Zn <sup>2+</sup>	15-100 ppm	Activator or component of many enzymes
Molybdenum (Mo)	MoO <sub>4</sub> <sup>2-</sup>	0.1-5.0 ppm	Required for nitrogen fixation and nitrate reduction
Boron (B)	B(OH) <sub>3</sub> or B(OH) <sub>4</sub> <sup>-</sup>	5-75 ppm	Influences Ca <sup>2+</sup> utilization, nucleic acid synthesis, and membrane integrity

C HOPKINS CaFe Mg B Mn CuZn Mo (Raven et al. 1992, *Biology of Plants*)

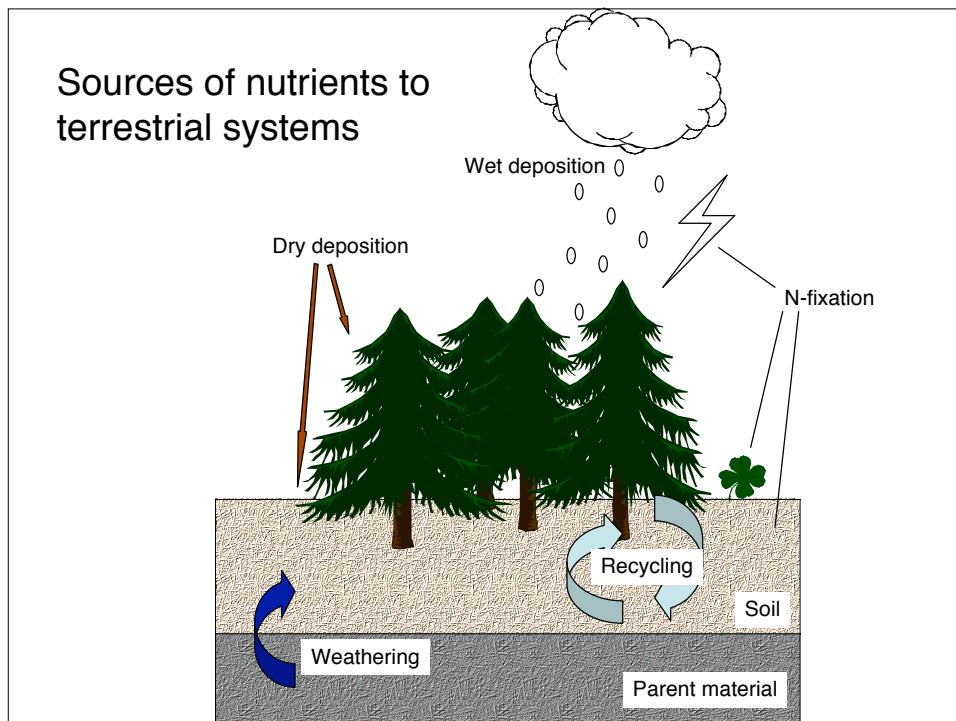
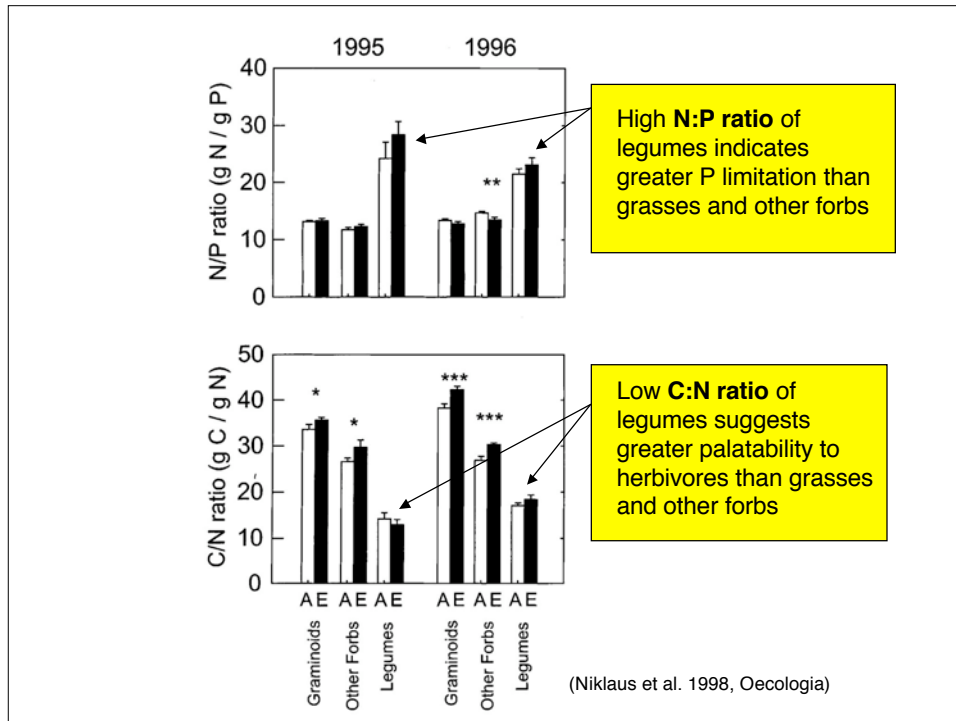
## Nutrient limitation of plant growth

- N limitation
  - most non-tropical terrestrial ecosystems; early 1° succession
- P limitation
  - highly weathered soils, including many tropical soils; calcareous soils
  - Different species within a community may be limited by different nutrients
  - Nutrient limitation usually demonstrated by response to fertilization

N:P < 14 N limitation  
 14 < N:P < 16 co-limitation  
 N:P > 16 P limitation  
 N:P = 10 optimal growth



(Aerts & Chapin 2000, *Adv. Ecol. Res.*)

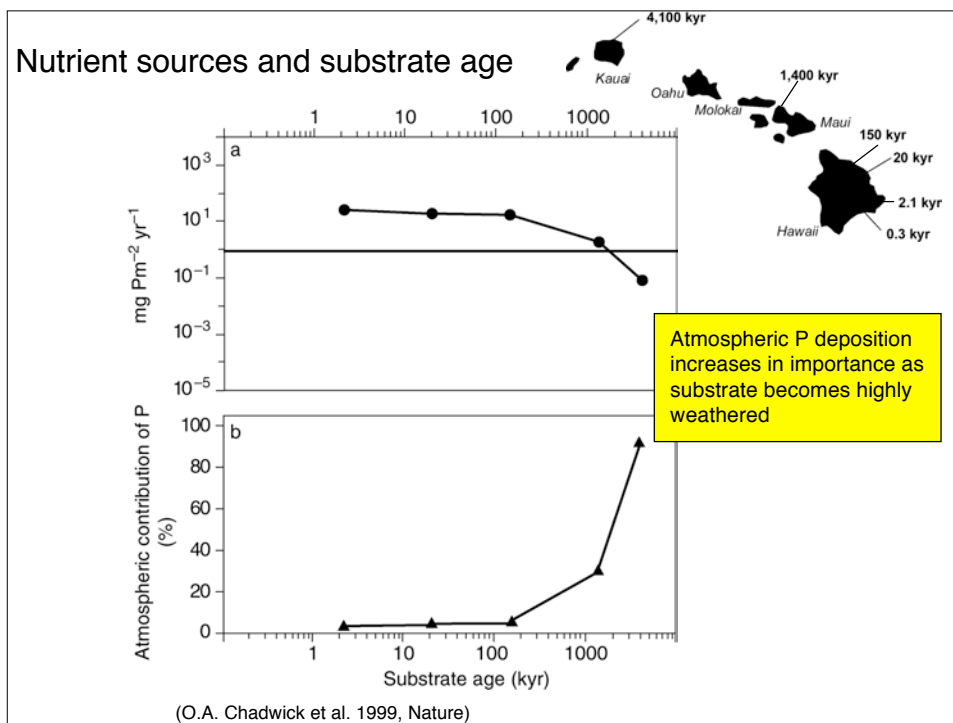
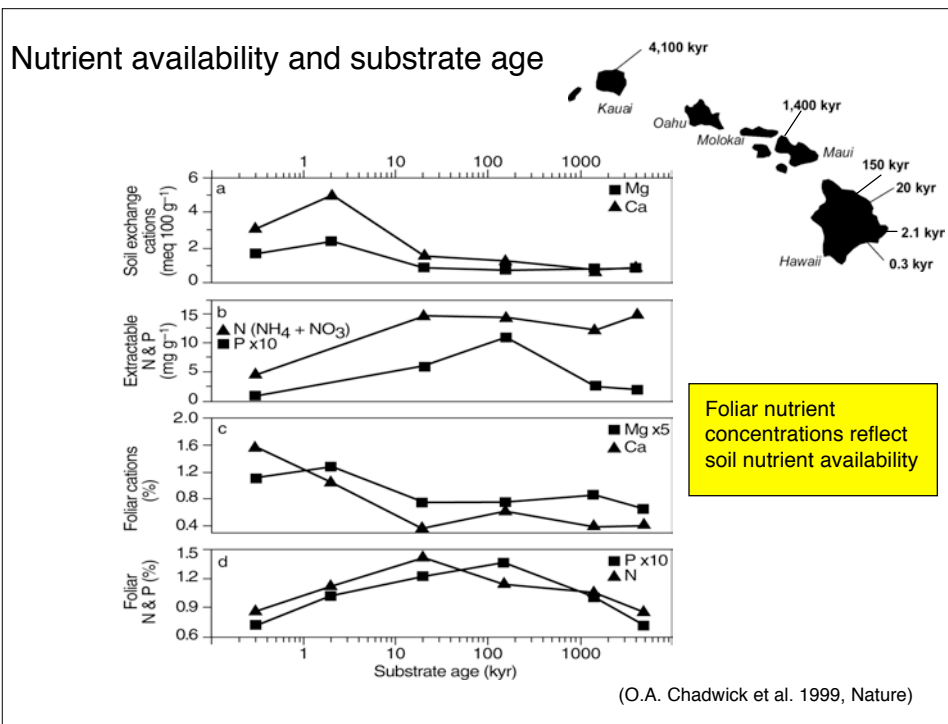


## Atmospheric sources of nutrients

- Rain, clouds, and fog (wet deposition)
  - Aerosols and gases are dissolved by rain or fog
- Dry deposition (sedimentation)
  - Gravitational deposition of particles; dust, soil and sea salts can provide important source of cations ( $\text{Ca}^{2+}$ ,  $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{Mg}^{2+}$ , P)
- Largest reservoir of N as  $\text{N}_2$  gas
  - $\text{N}_2$  fixed to  $\text{NH}_4^+$  by lightening ( $< 20 \text{ Tg yr}^{-1}$ ), biological N-fixation ( $140 \text{ Tg yr}^{-1}$ ), anthropogenic fixation ( $140 \text{ Tg yr}^{-1}$ )

## Weathering of parent material

- Weathering = f (parent material, climate, vegetation, topography, time)
- Parent material - chemical composition, texture
- Climate - temperature, precipitation; freeze-thaw
- Vegetation - root action; effects on pH
- Topography - erosion, movement of nutrients
- Time - time-dependant processes; nutrient depletion

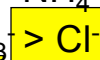
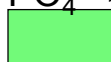


## Controls on litter decomposition

- Three major controls on decomposition:
  - climate > litter chemistry > soil organisms
- Climate most important determinant on global scale
  - microbial activity increases with temperature
  - intermediate moisture conditions maximize activity
- Within climate zone, litter chemistry strongest control
  - faster decomposition with higher nutrient content
  - litter chemistry correlated with site fertility

## Cation exchange complex

- Clay particles and humic substances in soil have electrically-charged surfaces, which attract and bind ions (adsorption)
- Net negative charge = more sites for cations than anions
- Tendency for adsorption varies:
  - cations:  $\text{Al}^{3+} > \text{Ca}^{2+} > \text{Mg}^{2+} > \text{NH}_4^+ > \text{K}^+ > \text{Na}^+$
  - anions:  $\text{PO}_4^{3-} > \text{SO}_4^{3-} > \text{NO}_3^- > \text{Cl}^-$

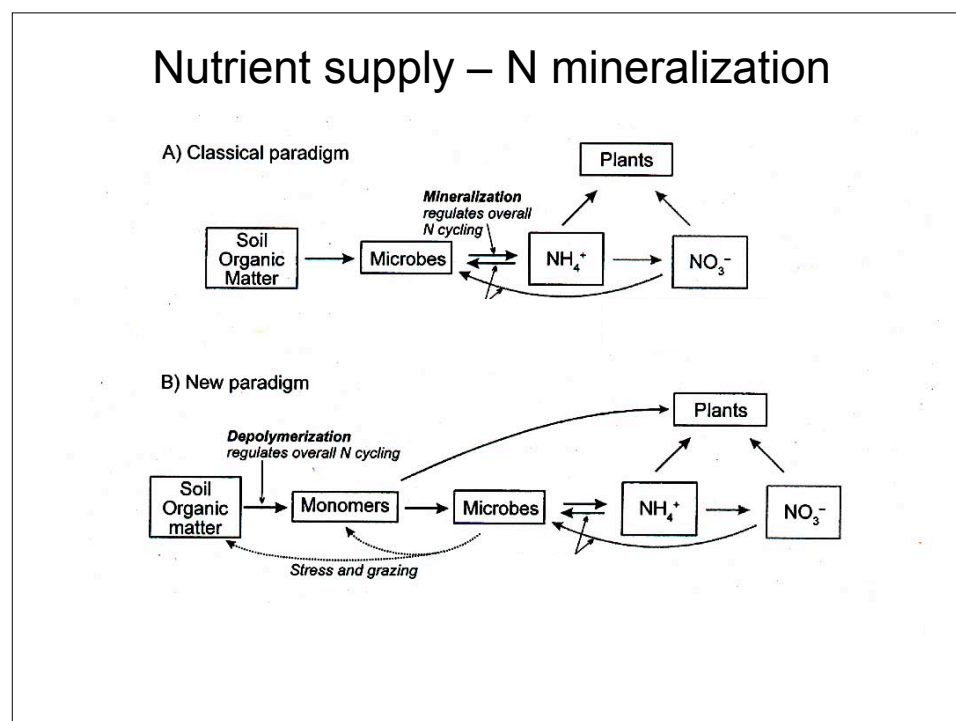


## Nutrient pools vs. rate of supply

Nitrogen dynamics in California annual grassland

	Soil pool (mg m <sup>-2</sup> )	Plant uptake (mg m <sup>-2</sup> d <sup>-1</sup> )	Microbial uptake (mg m <sup>-2</sup> d <sup>-1</sup> )	Total uptake (mg m <sup>-2</sup> d <sup>-1</sup> )
<b>February</b>				
NO <sub>3</sub> <sup>-</sup>	10	41	81	122
NH <sub>4</sub> <sup>+</sup>	640	81	426	507
<b>April</b>				
NO <sub>3</sub> <sup>-</sup>	51	83	146	229
NH <sub>4</sub> <sup>+</sup>	800	110	639	749

(Jackson et al. 1989. Soil Biology & Biochemistry)



## Ion movement by mass flow

- Transpiration stream
  - Rate of supply depends on transpiration rate/soil moisture and ion concentration in soil solution
- Wetting fronts
  - Highest nutrient concentrations at soil surface; infiltration of rain water can move nutrients to deeper soil
- Lateral movement in wet systems
  - Arctic tundra - permafrost causes lateral movement of water

## Nutrient supply – water availability

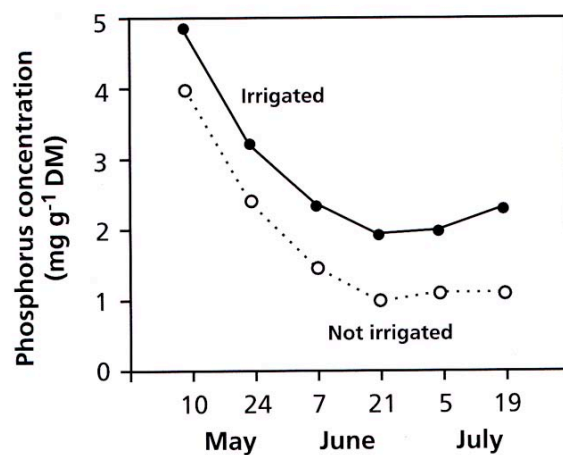


FIGURE 4. Phosphorus concentration in the shoots of *Hordeum vulgare* (barley) grown with or without irrigation (Chapin 1991). Copyright Academic Press.

## Ion movement by diffusion

$$D = D_1 \times \theta \times f \times 1/b$$

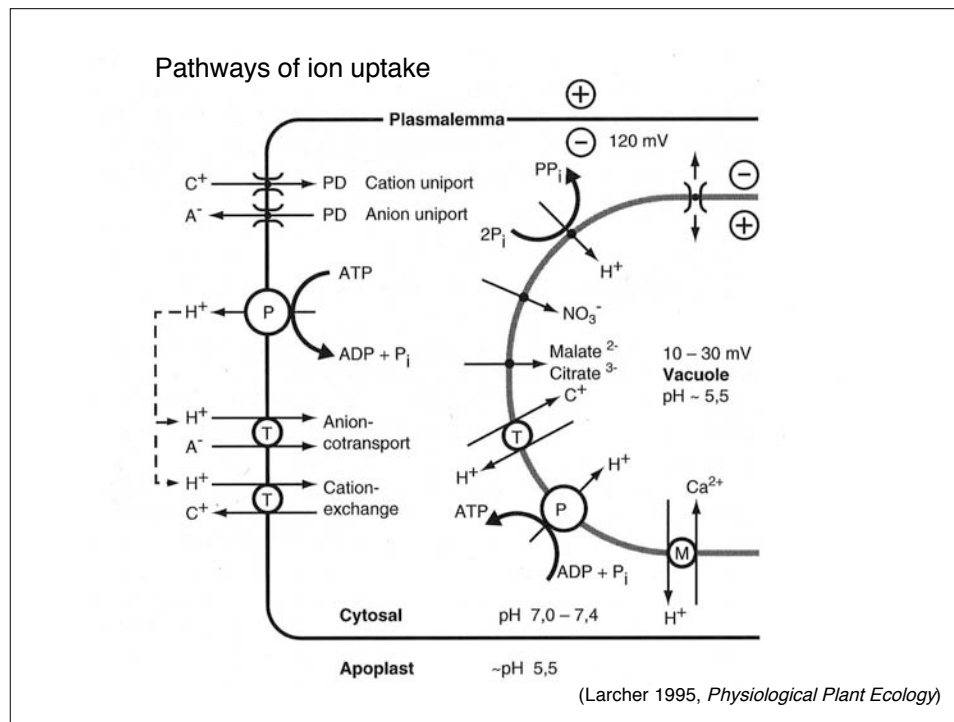
- where  $D_1$  = diffusion coefficient in free solution
- $\theta$  = volumetric water content of soil
- $f$  = an impedance factor
- $b$  = soil buffer capacity
- $f$  is the length of the diffusion pathway, which is a function of soil moisture and soil compaction
- Soil buffer capacity is determined by the supply of ions adsorbed on the Cation Exchange Complex (CEC)

$D_1$  for some soil solution ions ( $\text{m}^2 \text{s}^{-1}$ ):

$\text{NO}_3^-$   $110 \times 10^{-10}$ ;  $\text{K}^+$   $1-2810 \times 10^{-12}$ ;  $\text{H}_2\text{PO}_4^-$   $0.3 - 3.310 \times 10^{-13}$

## Ion uptake

- Concentration gradient across root surface does not favor ion diffusion into the root
- Cations diffuse along electrochemical potential gradient
  - Proton-pumping ATP-ase pumps  $\text{H}^+$  out of cell
- Anions must be actively transported against the electrochemical potential gradient



## Acclimation/adaptation to nutrient availability

- Uptake kinetics
- Root allocation and morphology
- Rhizosphere chemistry
- Growth strategies
- Symbioses

## Uptake kinetics

- Response to low nutrient supply
  - increase  $I_{\max}$  (maximum inflow rate; function of the abundance or specific activity of transport proteins)
  - induction of high-affinity transport system (carrier-mediated; has low  $I_{\max}$ )
  - requires investment of energy and proteins

## Root allocation and morphology

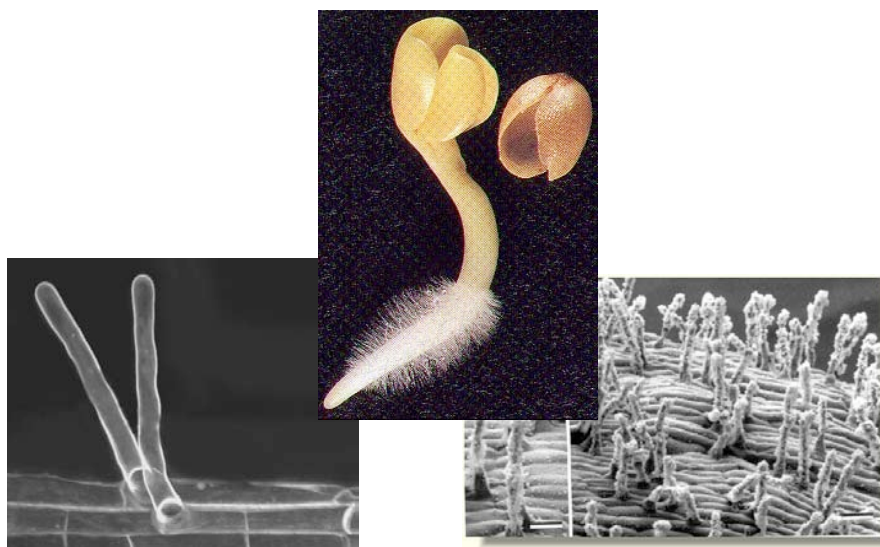
- Root mass ratio (% roots)
  - high RMR increases absorbing surface; largest investment
- Root hairs
  - greatly increase absorbing surface (high surface area to volume ratio); moderate investment
- Cluster roots
  - proteoid roots; associated with high rates of organic acid excretion

## Root mass ratio

N supply (mg/L)	Shoot (g/plant)	Root (g/plant)	S:R ratio	Root length (m)
0	0.24	0.38	0.63	4.7
21	0.75	0.84	0.89	6.2
42	1.34	1.30	1.03	6.8
105	2.40	1.97	1.25	8.1
210	4.49	2.89	1.55	10.2

(Corn – from Marschner 1986)

## Root hairs – increase absorptive surface (P)



## Non-mycorrhizal (Proteaceae) Cluster roots – P uptake



## Roots and spatial heterogeneity



Extreme example  
-Barley-

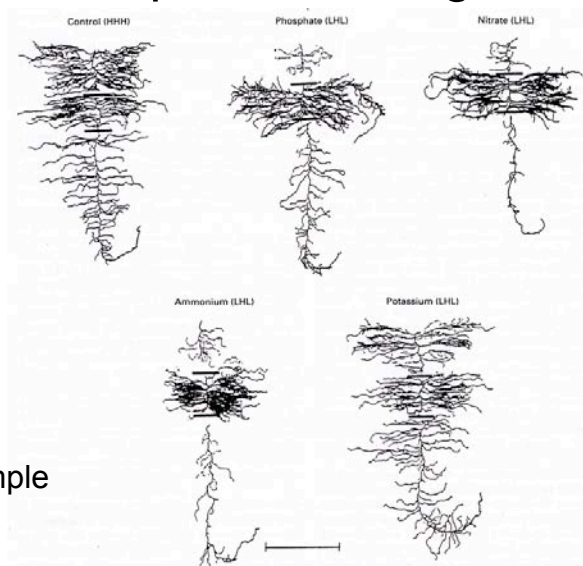
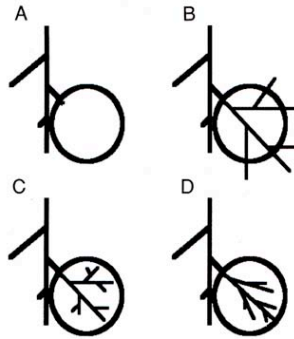


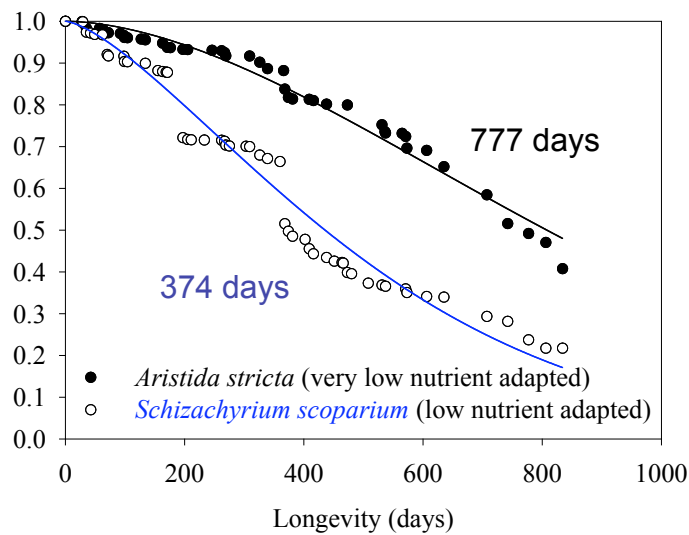
Figure 1. Single first-order axes of the seminal root system of *Hordeum vulgare* grown for 21 d in irrigated sand. All parts of the root systems of control plants were uniformly supplied with a concentrated nutrient solution (HHH treatment). In localized supply treatments (LHL), only the middle zones received the complete solution, the top and bottom zones being supplied with a solution deficient in either P,  $\text{NO}_3^-$ ,

## Roots and spatial heterogeneity



**Figure 6** Topological diagrams of model root systems to indicate the distinction between biomass allocation and architectural plasticity. An initial root system (A) may respond to a patch, indicated by a circle, by increased allocation of biomass with no change in architecture (B) or a change in architecture (C and D). Architectural changes may, for example, be a change in topology, leading to a less herringbone system, with no increase in allocation (C), or a change in geometry, such as shorter link lengths and narrower branching angles (D). Biomass allocation plasticity alone (B) is ineffective at localizing new growth within the patch.

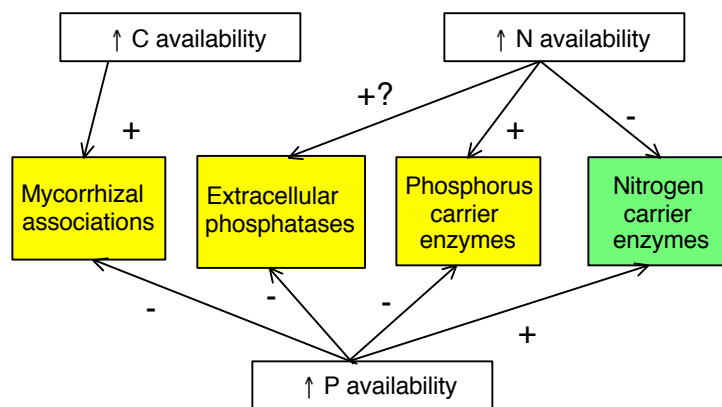
## Root lifespan



## Rhizosphere chemistry

- Excretion of  $H^+$  or organic acids reduces pH
  - increases availability of Zn, Mn, B, Mn, Fe
- Excretion of chelating agents
  - aids uptake of Fe and Zn; releases  $PO_4^{2-}$  bound in insoluble forms
- Excretion of phosphatases
  - cleave organic bonds to release  $PO_4^{2-}$
- Priming of rhizosphere mineralization
  - excretion of organic acids, carbohydrates, and amino acids stimulates microbial activity

## Resource availability and acquisition strategies



(Treseder & Vitousek 2000, Ecology)

## Nutrient recycling

**TABLE 1. Major sources of available nutrients that enter the soil.**

Nutrient	Source of plant nutrient (% of total)		
	Atmosphere	Weathering	Recycling
<b>Temperate forest</b>			
N	7	0	93
P	1	<10?	>89
K	2	10	88
Ca	4	31	65
<b>Arctic tundra</b>			
N	4	0	96
P	4	<1	96

*Source: Chapin 1991.*

## Plant growth strategies

- Characteristics of plants from low-nutrient environments
  - slow growth
  - low nutrient demand
  - long-lived tissues; C-based defenses
  - low tissue nutrient content; slow decomposition
- Characteristics of plants from high-nutrient environments
  - rapid growth
  - high nutrient demand
  - short-lived tissues; mobile N-based defenses
  - high tissue nutrient content; rapid decomposition

### Root allocation and morphology

- Root mass ratio (% roots)
  - high RMR increases absorbing surface
  - large investment
- Root hairs
  - greatly increases absorbing surface (high surface area to volume ratio)
  - moderate investment
- Get help!
  - mycorrhizae
  - nitrogen fixers

### Nutrient availability

- $\text{NO}_3^-$ 
  - highly mobile
  - leachable
- $\text{NH}_4^+$ 
  - less mobile
  - binds to SOM
- $\text{PO}_4^{-3}$ 
  - highly immobile
  - low solubility

### Mycorrhizal associations

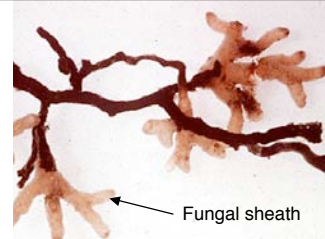
- occur in most plant species; low host specificity
- plant host provides C energy to fungus
- fungus increases effective absorptive surface
- particularly important in P nutrition; may also supply N and water



## Types of mycorrhizae

- Ectomycorrhizae

- fungus does not penetrate cortical cells
- characterized by fungal sheath
- primarily association between trees and basidio- or ascomycetes



- Vesicular/Arbuscular Mycorrhizae

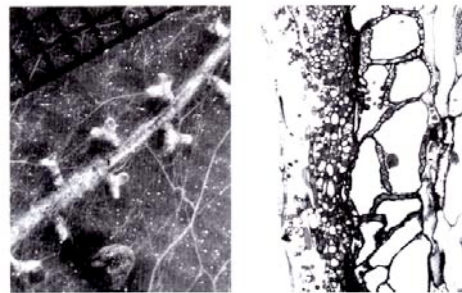
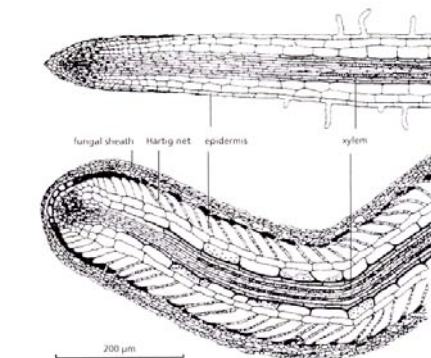
- fungus penetrates root cortical cells
- characterized by vesicles (storage), arbuscles (exchange)
- most frequently occur on herbaceous species, but also some trees

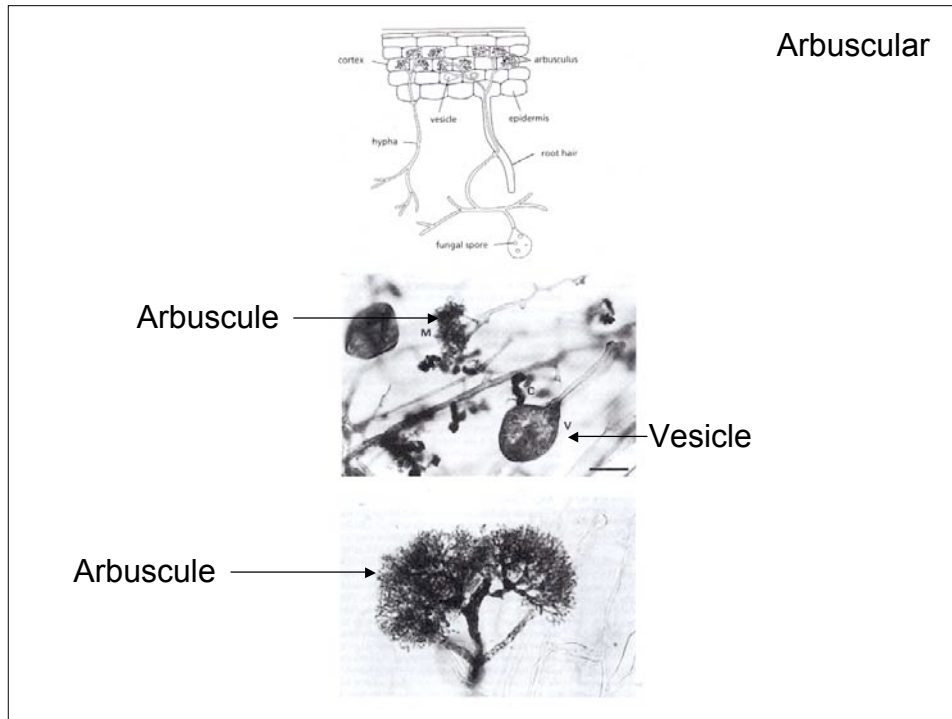
- Ericoid mycorrhizae

- similar to V/AM
- heath and tundra ecosystems



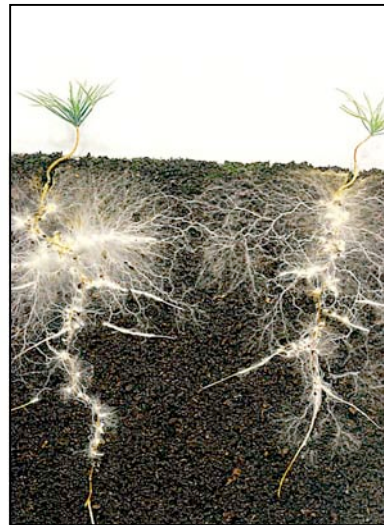
Ecto-





## Resource transfer via hyphal network

- Plant roots linked by fungal hyphae
  - interspecific and intraspecific
  - transfer of nutrients and carbon
  - net transfer appears to follow source-sink relationship
  - importance of transfer to plant nutrition not clear
  - transferred carbon may not be available to plant (retained in fungal storage structures)



(D. Read 1997, Nature)

## Plant benefits – surface area

TABLE 1. The length of mycorrhizal hyphae per unit colonized root length as measured for a number of plant species, infected with different mycorrhiza-forming fungal species.

Fungus	Host	Hyphal length (m cm <sup>-1</sup> root)
<i>Glomus mosseae</i>	onion	0.79–2.5
<i>Glomus mosseae</i>	onion	0.71
<i>Glomus macrocarpum</i>	onion	0.71
<i>Glomus microcarpum</i>	onion	0.71
<i>Glomus</i> sp.	clover	1.29
<i>Glomus</i> sp.	rye grass	1.36
<i>Glomus fasciculatum</i>	clover	2.50
<i>Glomus tenue</i>	clover	14.20
<i>Gigaspora calospora</i>	onion	0.71
<i>Gigaspora calospora</i>	clover	12.30
<i>Acaulospora laevis</i>	clover	10.55

Various authors, as cited in Smith & Gianinazzi-Pearson 1988.

## Symbiotic N-fixation

- reduction of N<sub>2</sub> to NH<sub>3</sub> catalyzed by the **nitrogenase** enzyme
- requires low O<sub>2</sub> environment; **leghemoglobin** controls nodule O<sub>2</sub>
- Nitrogenase consists of 2 proteins
  - Fe-S protein accepts e<sup>-</sup>, binds ATP
  - Fe-Mo protein binds N<sub>2</sub>
- energetically expensive means of acquiring N



## Symbiotic N-fixation

- restricted to more limited group of plants (legumes); high host specificity
- plant host provides C energy to bacterial N-fixer
- N-fixer provides available  $\text{NH}_4$  to plant

